N-Oxides and Related Compounds. Part 60.¹ Novel Thermal and Photochemical Rearrangements of *N*-Substituted 2-Pyridones ²

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Examples of four novel rearrangements of derivatives of 1-hydroxy-4,6-diphenyl-2-pyridone are reported : all involve N–O fission and formation of 3-substituted or both 3- and 5-substituted-4,6-diphenyl-2-pyridones. (a) $1-OCH_2CH_2R$ (R = vinyl or phenyl) compounds give $3-CH_2R$ (R = vinyl or phenyl) derivatives with elimination of CH₂O. (b) The 1-octyloxy-compound gives the 3-octyloxy-derivative by simple transposition. (c) 1-Acyloxy-compounds form the corresponding 3- and 5-acyloxy-2-pyridones. (d) 1-Imidoyloxy-compounds yield the rearranged 3- and 5-amido-2-pyridones. The mechanisms probably all involve homolytic N–O fission.

OUR studies ³ on the thermolysis and photolysis of 1alkoxy-4,6-diphenyl-2-pyridones to give aldehydes have shown that, whereas benzaldehydes are formed without appreciable amounts of side products, the aliphatic aldehydes obtained from this route are accompanied not only by alcohols but also by by-products derived from rearrangements. We have now elucidated the structures of these by-products and the courses of these and related rearrangements (which increase significantly the range of those known for heteroaromatic *N*oxides ^{4,5}).

Rearrangements with Loss of Formaldehyde.—Thermolysis of the phenethoxypyridone (1) at 220 °C yields [in addition to phenylacetaldehyde (49%) and 4,6-diphenyl-2-pyridone 36] 26% of 3-benzyl-4,6-diphenyl-2-pyridone (3). The i.r. [amide v(C=O) 1 630; v(NH) 2 800-3 100 cm⁻¹] and ¹H n.m.r. spectra (2 H, s at δ 3.7; 1 H, s at δ 6.4; 15 H, m at δ 6.8—7.7; and 1 H, broad s at δ 12.0) were similar to those of 4,6-diphenyl-2-pyridone. These, together with ¹³C n.m.r. and mass spectra and analytical data, indicated the structure (3) or the isomeric 5-benzyl compound (5): distinction was achieved by elimination of the oxygen atom. In model reactions, 4.6-diphenyl-2-pyridone was converted by POCl₂ at 200 °C into the 2-chloropyridine (6) (84%) which on hydrogenation at 20 °C and 760 mmHg gave 2,4diphenylpyridine (7) (80%), for which 6-H displayed a clear doublet at 8 8.75 (/ 10 Hz) in the ¹H n.m.r. spectrum. Under the same reaction conditions we converted the benzyl derivative (3) successively into the chlorocompound (8) (96%) and 2,4-diphenyl-5-benzylpyridine (9) in which the 2-proton signal was a singlet at δ 8.8. This eliminates structure (5) and strongly supports (3) for the original benzyl derivative.

Similar thermal rearrangements with loss of formaldehyde occur in the but-3-enyloxy series. Thermolysis at 230 °C of 1-(but-3-enyloxy)-4,6-diphenyl-2pyridone (2) gave [in addition to crotonaldehyde $(50\%)^3$] 3-allyl-4,6-diphenyl-2-pyridone (4) (33%). The i.r. [amide ν (C=O), 1 610; ν (NH), 2 800—3 200 cm⁻¹] and ¹H n.m.r. data (see Experimental section) together with analytical data narrowed the structure to the 3- and 5-

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allyl isomers; the former is considered more likely particularly by analogy with the benzyl analogue.

We also converted 1-phenethyloxy-4,6-diphenyl-2pyridone into phenylacetaldehyde (30%) by photolysis, details of which are given elsewhere 3b: now, using chromatographic separation of the products we find



rearrangement with elimination of formaldehyde also occurred to give (3) (14%).

Rearrangement to Ring-substituted Isomeric Products.— We have reported 3b that 1-n-octyloxy-4,6-diphenyl-2pyridone (11) on thermolysis gives n-octaldehyde (37%) and n-octanol (13%); photolysis gives the same products in yields of 47% and 23%, respectively. We have now



found that preparative t.l.c. of the pyrolysis residue gives a third product in 5% yield, shown analytically to be an isomer of the starting material: the component was also isolated from the photolysis residue in 2% yield. The i.r. [amide ν (C=O), 1 600—1 625; ν (NH) 2 700–3 100; v(C-O-C) 1 230 and 1 040 cm⁻¹] and ¹H n.m.r. spectra [δ 1.6 (15 H, m), 4.0 (2 H, t), 6.5; (1 H, s), 7.3–7.9 (10 H, m), 12.7 (1 H, br s)] indicate either 3- (13) or 5-n-octyloxy-4,6-diphenyl-2-pyridone as the structure: the former is more likely.

Photolysis of 1-(5-nitro-2-pyridyl)-4,6-diphenyl-2-pyridone (12) in dry benzene gave the 3-pyridylpyridone (14) (50%) as a brown crystalline solid, insufficiently stable to permit isolation as the pure material. Assignment was based on i.r. [amide v(C=O) 1 640, v(N-H) 2 500—3 500, $v(NO_2)$ 1 350, v(C-O) 1 250 cm⁻¹] and ¹H n.m.r. spectral data: in particular the latter showed absorptions characteristic of a 2,5-disubstituted pyridine and a trisubstituted 2-pyridone. Again the 3-isomer is considered more likely than the 5-isomer by analogy with the preceding rearrangements.

Photochemical Rearrangements of 1-Aroyloxy- and 1-Acyloxy-4,6-diphenyl-2-pyridones.— 1-Acyloxy-2quinolones were reported some years ago.⁶ Paquette ^{7,8} prepared 1-acyloxy-2-pyridones by reaction of acyl chlorides with 2-ethoxypyridine 1-oxide and utilised them as acylating agents.^{8,9} They have also been prepared using thallium salts.¹⁰

We prepared the 1-acyloxy-4,6-diphenyl-2-pyridones (15a-e) from the sodium salt (10) and the corresponding acid chlorides: the structures are supported by spectral data (see Experimental section), particularly the ester v(C=O) absorption at 1.780-1.815 cm⁻¹ and the amide v(C=O) absorption at *ca*. 1 670 cm⁻¹. The 1-acyloxy-compounds are sensitive to water, especially the 1-acetoxy (15a) and 1-phenylacetoxy (15b) derivatives.

Pyrolysis of the 1-acyloxy-derivatives (15a, b, d, and e) gave merely tar and some of the corresponding acid. We photolysed our compounds (15d and e) in acetonitrile to minimise solvent participation. At 253.7 nm, (15e), which shows u.v. $\lambda_{max.}$ at 250 and 330 nm, gave a mixture of products from which only 4,6diphenyl-2-pyridone was isolated, but irradiation at 350



nm gave, after 36 h, rearranged ester (16) (12%) together with 4,6-diphenyl-2-pyridone (15%) and benzoic acid (2%).

Structure (16) was supported by ester ν (C=O) 1 740, pyridone ν (C=O) 1 650, and pyridone ν (N-H) 2 600— 3 000 cm⁻¹ absorptions. In the ¹H n.m.r. spectrum the AB quartet of the starting material, for the 3-H and 5-H protons, had collapsed to a one-proton singlet at δ 6.6, but the signal for the ortho protons of the benzoyl group remained as a doublet of δ 7.8. The mass spectrum disclosed a molecular ion peak at m/e 369, and a base peak at m/e 105 for PhCO⁺. Hydrolysis of (16) in 20% aqueous sodium hydroxide gave benzoic acid and the hydroxypyridone (17) [v(OH) 2 500-3 500, pyridone v(C=O) 1 640 cm⁻¹; 3-H singlet at δ 7.25 with other aromatic protons near δ 7.65].

Irradiation at 350 nm of 4,6-diphenyl-1-(p-toluoyloxy)-2-pyridone (15d) in dry acetonitrile for 38 h and separation by column chromatography gave both the rearranged 5- (18) (10%) and 3-p-toluoyloxy-isomers (19) (16%). p-Toluic acid (12%), 4,6-diphenyl-2-pyridone (35%), and other minor unidentified products were also obtained.

The rearranged esters (18, 19) had similar physical and spectral properties to those of (16) obtained earlier: v(NH) 2500-3100, v(C=O) 1740 and 1650-1660cm⁻¹. The ¹H n.m.r. spectrum showed the characteristic AB pattern and the Me singlet of the *p*-toluoyl group, with a 3-H singlet at δ 6.6 (CDCl₃) for the 5-*p*toluoyloxy-isomer (18) and a 5-H absorption for the 3*p*-toluoyloxy-isomer (19) (CDCl₃) at δ 6.4 (obscured by the aromatic peaks in CF₃CO₂H). Mass spectra did not distinguish the two isomers but provided further evidence for the proposed structures [molecular ion m/e**381**; base peaks m/e 119 (MeC₆H₄CO⁺) and 104, (PhC= NH⁺)].





The two toluoyloxy-isomers (18) and (19) were separately hydrolysed by sodium hydroxide into p-toluic acid and the corresponding hydroxypyridones (17) and (20), the first identical to that obtained from ester (16). The other hydroxypyridone (20) was characterised by elemental analysis and i.r. $[v_{max}, 3\ 350\ (OH), 3\ 060\ (NH),$ and 1 630 cm⁻¹ (C=O) and ¹H n.m.r. spectroscopy [8 6.6 (1 H, s, 5-H), 7.2--7.8 (10 H, m), and 12.5 (exchanged by D_2O]. Mass spectral fragmentation patterns are characterised by loss of H (m/e 262) followed by sequential loss of two CO units (m/e 234 and 206). For (17) a competing fragmentation involves loss of CO from the parent ion. Both spectra exhibit m/e 104 assigned to PhC=NH. Compared with the ¹³C n.m.r. spectrum for 4.6-diphenyl-2-pyridone, those for (17) and (20) each showed one ring peak (C-3 or C-5) shifted downfield by the attached hydroxy-group and a singlet in the offresonance-decoupled spectrum; the other peak remained approximately at the same chemical shift, remaining as a doublet on off-resonance decoupling. However, due to the similar chemical shifts of carbon atoms C-3 and C-5 (113.7 and 104.2 p.p.m.) in 4,6-diphenyl-2-pyridone, it was not possible to determine unambiguously which peak had shifted, and thus to identify (17) and (20).

To differentiate the 3- and 5-substituted series, pyridone (19) was reduced to the pyridine (22). Phosphoryl chloride at 200 °C gave 2-chloro-4,6-diphenyl-3-(p-toluoyloxy)pyridine (21) (28%) and p-toluamide (47%) (after aqueous ammonia work-up) indicating appreciable ester cleavage. Hydrogenation of the chloropyridine (21) both removed the chlorine and cleaved the ester to give 3-hydroxy-4,6-diphenylpyridine (22) (89%) with the 2-H singlet absorbing at δ 8.5. This enabled structural assignment of the pyridone (19) and by extension those of (16), (17), (18), and (20).

The 5-toluoyloxy-isomer (18) has a lower u.v. absorption maximum (327 nm) than the 3-toluoyloxy-isomer (19) (333 nm), consistent with the structural assignments; the decrease in λ_{max} is attributed to the greater steric hindrance in the 5-toluoyloxy-isomer (18).

Photochemical Rearrangements of 1-Benzimidoyloxy-4,6-diphenyl-2-pyridones.— 4,6-Diphenyl-1-(N-phenylbenzimidoyloxy)-2-pyridone (24a) (88%) was prepared from N-phenylbenzimidoyl chloride (23a) ¹¹ and the salt (10) in dry acetonitrile at 20 °C. Similarly the 1-[N-(pmethoxyphenyl)-o-toluimidoyloxy]- (24b) (93%) and 1-[N-(p-tolylbenzimidoyloxy)]-4,6-diphenyl-2-pyridones (24c) (95%) were prepared from N-(p-methoxyphenyl)o-toluimidoyl chloride (23b) and N-(p-toluoyl)benzimidoyl chloride (23c),¹² themselves obtained from the corresponding amides and thionyl chloride.

$$Ar^1CCl = NAr^2$$

Each 1-benzimidoyloxypyridone (24a, b, and c) showed in the ¹H n.m.r. spectrum protons 3-H and 5-H as a pair of doublets near δ 6.45 and δ 6.95 (*J* 2.5 Hz) and the two protons *ortho* to the benzimidoyl group near δ 6.5 (upfield from the remaining aromatic protons) as a doublet (*J* 8 Hz) or in the case of the *p*-methoxyphenyl derivative (24b) contributing to a four-proton singlet, coinciding with the adjacent *meta* protons. The i.r. spectra all showed the pyridone ν (C=O) 1 660 and ν (C=N) 1 600 cm⁻¹ absorptions.

Pyrolysis of the 1-(N-phenylbenzimidoyloxy)-2-pyridone (24a) at 130 °C and 0.5 mmHg gave a mixture from which only benzanilide was identified. The u.v. spectra show that the benzimidoyloxy-derivatives absorb strongly near 250 nm and more weakly near 350 nm. Photolysis of (24a) in dry acetonitrile at 253.7 nm gave only a little rearranged product (26). Irradiation for 46 h at 350 nm in dry acetonitrile was more successful. 4,4'-Dibenzamidobiphenyl (25) (1.7%) precipitated from the photolysis mixture; the structure follows from elemental analysis, i.r. [amide v(NH) 3 240 and amide ν (C=O) 1 645 cm⁻¹] and ¹H n.m.r. spectra (18 aromatic protons, 2 NH protons), m.p. comparison [m.p. >335 °C (lit.,¹³ 352 °C)], mass spectral data [molecular ion peak, m/e 392; base peak, 105 (PhCO⁺); 77 (Ph⁺); 287 $(PhCONH=C_{6}H_{4}=C_{6}H_{4}=NH)$], and its insolubility in most solvents.13,*



Column chromatography and fractional crystallisation gave additional compounds from the 350 nm photolysis, viz. two rearranged benzanilidopyridones (26) (13.5%) and (28) (4%), 4,6-diphenyl-2-pyridone (18.5%), benzanilide (4%), two unknowns in moderate quantities (ca. 10% by weight), and several unidentified minor components.

The ¹H n.m.r. and i.r. spectra of the rearranged pyridones (see Experimental section) support the proposed structures. Satisfactory elemental analysis and mass spectrum [base and molecular ion peak, m/e 442; m/e 337, loss of PhCO; 309, further loss of CO; 105, PhCO⁺; 104, PhC=NH⁺; 77, Ph⁺] were obtained for the 3-substituted isomer (28). After eight recrystallisations a 2% error in carbon analysis was still found for isomer (26); a mass spectrum was unobtainable because of non-volatility (10⁻⁶ mmHg at 350 °C). However, hydrolysis of this isomer (26) in 70% sulphuric acid gave benzoic acid and 5-anilino-4,6-diphenyl-2-pyridone (27) [v(NH) 3 440 and 3 330, v(NC=O) 1 640 cm⁻¹], for

^{*} The m.p. evidence for (27), though equivocal, precludes the 2,2'-, 2,4'-, and 3,4'-isomers which have lower m.p.s (ref. 13; R. B. Carlin and E. A. Swakon, J. Amer. Chem. Soc., 1955, 77, 966; R. B. Carlin and W. O. Forshey jun., J. Amer. Chem. Soc., 1950, 72, 793). The unknown 2,3'- and 3,3'-isomers cannot be excluded but are considered less likely.

which ¹³C n.m.r. shows a similar C-3 signal to that observed for 4,6-diphenyl-2-pyridone but a lack from the usual position of a signal for C-5. The 3-anilidopyridone (28) was hydrolysed by sulphuric acid to benzoic acid, the expected 3-anilino-2-pyridone (29) not being obtained pure.

To obtain rearrangement products with more informative ¹H n.m.r. spectra, 1-[N-(p-methoxyphenyl)-o-toluimidoyloxy]-4,6-diphenyl-2-pyridone (24b) was irradiated at 350 nm in dry acetonitrile for 48 h. Column chromatography, fractional crystallisation, and preparative t.l.c. separated 4,6-diphenyl-2-pyridone (25%) and the amide (31) (40%) (both identified by comparison with authentic samples), and a rearranged pyridone (30) (15%).

The ¹H n.m.r. spectrum of (30) displayed methoxy, methyl, phenyl, and tolyl peaks as expected; the AB quartet for the *p*-methoxyphenyl ring appeared near δ 6.0, and the 3-H singlet at δ 6.5. Mass spectral fragmentations include initial loss of 2-MeC₆H₄CO (*m/e* 367) followed by loss of CO (*m/e* 339). A peak for *m/e* 104, PhC=NH⁺ is also present.





Attempted hydrolysis of (30) in 70% sulphuric acid gave o-toluic acid with intractable tar: lower concentrations of sulphuric acid or refluxing aqueous sodium hydroxide or hydrochloric acid gave recovered starting material. Mixed hydrochloric acid and acetic acid in a sealed tube yielded the 5-aminopyridone (32), an assignment supported by mass spectrometry [molecular ion peak, m/e 262; m/e 261 ($M^+ - H$), 104, (PhC=NH⁺), 77 (Ph⁺)], i.r. [v_{max} . 3 430 (NH₂), 3 300 (NH₂), and 1 655 cm⁻¹ (C=O)], and ¹H n.m.r. spectral data [δ 2.2 (NH₂), 6.5 (3-H), 7.5 (2 × Ph)] although it was obtained in insufficient quantity to purify for elemental analysis.

To obtain rearrangement products which would undergo hydrolysis more smoothly, 4,6-diphenyl-1-[N-(p-tolyl)benzimidoyloxy]-2-pyridone (24c) was photolysed at 350 nm. Differential solubility and column chromatography afforded 4,6-diphenyl-2-pyridone (12%) and benzo-p-toluidide (14%) (both identified by comparison with authentic samples), two rearranged pyridones (33) (16%) and (35) (18%), and a compound (14%) which analysed as C₂₁H₁₉O₂. Two alternative possible structures (39) and (40) are proposed for the compound $(C_{21}H_{19}O_2)$ on the basis of i.r. $[\nu(C=O) \ 1\ 655\ cm^{-1}]$, ¹H n.m.r. $[\delta\ 1.6\ (3\ H, s, Me), 2.2\ (3\ H, s, Me), 6.2\ (2\ H, d, J\ 10\ Hz)$, and 7.8—8.4 (11 H, m)], and mass spectral data [molecular ion peak, $m/e\ 317$; $m/e\ 211$ (loss of MeC₆H₄O); base peak, $m/e\ 107$, (MeC₆H₄O⁺); $m/e\ 105$, (PhCO⁺); and 77, (Ph⁺)].



The two benzotoluidinopyridones (33) and (35) gave satisfactory elemental analyses and showed the expected i.r. spectra (see Experimental section). The ¹H n.m.r. spectra both had methyl singlets near δ 2.1. The 3-substituted isomer (33) showed singlets for 5-H at δ 6.5 (in CDCl₃) and NH at δ 9.4, while the 5-substituted isomer (35) showed a singlet for 3-H at δ 6.6 (in CDCl₃; poor solubility) or at δ 6.9 (in CF₃CO₂H). The mass spectra of both isomers had the expected molecular ion (m/e 456) and base (m/e 105, PhCO⁺) peaks.

The two rearranged pyridones were each hydrolysed by 70% sulphuric acid to benzoic acid and the corresponding toluidinopyridones (34) and (36). The proposed structures were supported by analyses and spectra: v(NH) 3 330 and 3 400 cm⁻¹ (34), or 3 320 and 3 440 cm⁻¹ (36); pyridone ν (C=O) 1 625 cm⁻¹ (34) or 1 650 cm⁻¹ (36). The pyridone ring ¹H n.m.r. peaks overlapped with the p-toluidino-absorption but 3-H was distinguished as a singlet at δ 6.3 for (36) [in (CD₃)₂SO] and 5-H as a singlet at δ 6.5 for (34) (in CDCl₃). The mass spectral fragmentation patterns both include loss of H $(m/e \ 351)$. This precedes loss of CO for (36) but competes with it for (34). Both spectra include m/c104 found elsewhere in this series. Comparison of the ¹³C n.m.r. spectra with that of 4,6-diphenyl-2-pyridone is complicated by the presence of the toluidino-group; however the C-5 doublet occurs at 108.5 p.p.m. in the off-resonance decoupled spectrum of the 3-substituted isomer (34) (in CDCl₃), another doublet at 118.5 p.p.m.

being assigned to the *ortho* carbons of the toluidinogroup. The 5-substituted isomer (36) (in trifluoroacetic acid-CDCl₃) showed the C-3 signal near the aryl carbon absorptions.

Again physical methods failed to differentiate clearly between the 3- and 5-substituted pyridones (33) and (35). The appearance of C-3 downfield from C-5 in the ¹³C n.m.r. spectra supports the suggested structural assignment; however, changes in chemical shift with solvent render this evidence uncertain. The problem was resolved chemically. The pyridone (35) was reduced to the corresponding pyridine (38) via initial reaction with phosphoryl chloride to give the 2-chloropyridine (37) (32%), characterised by analysis and spectra (see Experimental section). The pyridine (38) was identified by analysis and spectra: in the ¹H n.m.r. spectrum the 2-H proton appeared as a doublet ($J \in Hz$) at $\delta \otimes 8.8$, showing that the rearrangement had located the benzoyltoluidino-group at the 5-position and confirming all the structural assignments in this series.

U.v. spectra of the amido- and amino-pyridones (26)— (30) and (33)—(36) support the structures assigned above (see Table). The 5-substituted compounds have a lower and weaker absorption maximum than the corresponding 3-substituted isomers, attributed to the greater steric hindrance in the 5-isomers. The differences in wavelength between the maxima of the toluidino- and Nbenzoyltoluidino-isomers are greater than the corresponding differences between the maxima of the anilino- and benzanilido-isomers, due to the greater bulk of the toluidino- and N-benzoyltoluidino-substituents.

U.v. spectral data of amino- and amido-pyridones (in CHCl_a)

	•					
	3-Isomer			5-Isomer		
	Cmpd.	$\lambda_{max.}$		Cmpd.	λ_{max} .	
Substituent	no.	(nm)	10 ⁻³ ε	no.	(nm)	10 ⁻³ ε
Benzanilido	(28)	354	19.9	(26)	<340 ª	;
Anilino	(29)	355	b	(27)	348	b
N-(o-Toluoyl)-p-anisidino			(30)	327	7.7	
N-Benzoyl-p-toluidino	(33)	351	15.0	(35)	332	9.0
Toluidino	(34)	347	12.7	(36)	318	9.0
^a Overlapped by ab	sorption	atλma	. 293 nn	n. ⁰ No	t comp	letely
dissolved.	-				1	2

Pyridone Rearrangement Mechanisms.—There are few close analogies in pyridine 1-oxide chemistry to the rearrangements presented above. The photochemistry of derivatives of 1-hydroxy-2-pyridones has received little previous attention. Furrer ¹⁴ photolysed 1-alkoxy-4,6dimethyl-2-pyridones, but found only cleavage to give aldehydes, and in one instance the alcohol. No rearrangement products were observed although the *N*piperidino-derivative (41) gave the 3-substituted pyridone (42) (10%) together with 4,6-dimethyl-2-pyridone (56%). Taylor *et al.*¹⁵ photolysed 1-aroyloxy-2-pyridones into arylcarboxylic acids and arylbenzenes (on trapping with benzene); no rearrangement was observed.

2-Picoline 1-oxide and acetic anhydride give an initial N-acetoxypicolinium salt (43). This loses a proton to produce the anhydro-base (44) which undergoes intramolecular N-O bond cleavage and recombin-

ation to give (46)—(48): evidence for both heterolytic cleavage to give ion pairs ¹⁶ and homolytic cleavage to give radical pairs ^{17,18} has been presented. The anhydrobase (44) is a methylene analogue of a 1-acyloxy-2pyridone: rearrangement of (44) in part to 3- and 5acetoxypicoline (46, 48) is analogous to the rearrangements outlined above to 3- and 5-acyloxypyridones.



Such rearrangements of *N*-acyloxy-groups to the side chain or to a β -position occurs with many heterocyclic *N*-oxides: those of 1-methylisoquinoline,¹⁹ 4-lepidine,^{19,20} and 4-picoline ^{21,22} give higher yields of β -ringsubstituted products. Participation of intermolecular processes in the rearrangement of the anhydro-bases of 4-picoline ^{18,22,23} and lepidine ²⁰ has been suggested by Oae and his co-workers.

Abramovitch and his co-workers found rearrangements of heteroaromatic *N*-oxides on treatment with aryl isocyanates, imidoyl chlorides, nitrilium salts, activated acetylenes, and benzyne.⁵ Thus, an *N*-oxide blocked at the α -positions, such as lutidine 1-oxide, reacts with *N*-phenylbenzimidoyl chloride to give, among other products, the 3-imidate (51).²⁴ Initial nucleophilic attack by the *N*-oxide oxygen is followed by intramolecular addition, the 1,2-dihydropyridine intermediate (49) undergoing a [1,5] sigmatropic shift to the



2,3-dihydro-isomer (50) which then aromatises into the 3-imidate (51). Aryloxypyridinium salts such as (52) rearrange in the presence of nucleophiles to 3-(o-hydroxyphenyl)pyridines (53); 25 the isolation of small amounts of pyridine and p-nitrophenol suggests the involvement of radicals.

Homolytic cleavage of N-O bonds has been observed by e.s.r. in the homolytic cleavage of the acetoxyimine (54) ²⁶ and in the rearrangement of the 1-(4-pyridyloxy)-4-pyridone.²⁷

Oae *et al.*²⁸ rearranged the arenesulphonoxy-groups from the 2- to the 4-position in 1-isoquinolone: the main pathway was *via* solvent-separated ion pairs, with some participation by an oxygen-bridged ion pair.

The thermal decomposition of 1-alkoxy-4,6diphenyl-2-pyridones leading to aldehydes, alcohols, and 3-substituted pyridones is, we believe, the result of two competing pathways. Aldehyde formation is rationalised in terms of a six-centre transition state involving proton abstraction by the amide carbonyl whereas the other products result from homolytic cleavage of the N-O bond to give an alkoxy-radical. The latter either abstracts a hydrogen atom or fragments to an alkyl





radical and a carbonyl compound, the relative importance of the two modes being a function of the alkyl radical stability.²⁹

1-Benzyloxypyridones give high yields of aldehydes consistent with the ease of abstraction of the benzylic proton. For other 1-alkoxy-derivatives the radical pathway competes. The primary n-octyloxy-radical recombines intact at the 3-position of the pyridone nucleus but in contrast the phenethyloxy- and but-3enyloxy-radicals first fragment with loss of formaldehyde. The resultant conjugated radical subsequently reacts with the pyridonyl radical, again at the 3-position.*

In the analogous photochemical reactions we propose excitation of the pyridone carbonyl function. Aldehyde formation arises from abstraction of a benzylic proton and the rearrangement products again arise from the alternative N-O cleavage pathway. The rearrangement of the pyridyloxy-pyridone (12) may involve a similar homolytic cleavage or a [1,5]sigmatropic shift.

The rearrangement of the 1-imidoyloxy-compounds with cleavage of an N-O bond and formation of a C-N rather than C-O bond demonstrates that the process does not follow a sequence of the type (49)—(51). Direct [1,5] and [1,3] sigmatropic shifts (to give the 3and 5-substituted products respectively) are precluded for the same reason. Concerted rearrangements (55) to (56) and (57) to (58), though attractive, seem unlikely



on steric grounds and do not account for the observed by-products, e.g. (25). We therefore favour N–O cleavage followed by recombination [*i.e.* mechanisms analogous to those indicated by the conversion (44) into (46)—(48)] of which the free-radical pathway is preferred over the ion-pair mechanism, bearing in mind the reaction conditions and instability of a pyridone cation. The rearrangements of the 1-acyloxy-pyridones appear to be analogous.

EXPERIMENTAL

All melting points were determined using a Reichert hot stage microscope. I.r. spectra were recorded on Perkin-Elmer 237, 257, or 297 spectrophotometers. ¹H N.m.r. spectra were recorded either on a Varian HA 100 spectrometer at 100 MHz or on a Perkin-Elmer R12 spectrometer at 60 MHz. Mass spectra were obtained either on a Hitachi-Perkin-Elmer RMU-6E or A.E.I. MS9 mass spectrometers. ¹³C N.m.r. spectra were recorded on a JEOL FX 100 spectrometer. Preparative thick layer chromatography was carried out on Kieselgel 60PF₂₅₄ silica gel. Except where stated, column chromatography was carried out on alumina (UG 01). Dry or purified acetonitrile refers to drying over MgSO4, followed by reflux over P₂O₅, distillation, and storage over molecular sieves. Dry diethyl ether and dry light petroleum refer to material dried over sodium wire. Dry ethyl acetate refers to material washed with aqueous sodium hydrogencarbonate (5%), and saturated aqueous sodium chloride, dried over MgSO₄, distilled, and stored over molecular sieves.

Except where stated, photochemical reactions were performed in a quartz glass vessel using a Rayonet photochemical reactor of sixteen tubular u.v. sources mounted around the inner surface of a polished cylindrical reflector.

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^{*} In response to a referee's comment we note that, in principle, aldehyde formation could derive from decomposition of alkoxyradicals implying that all products originate from initial homolytic cleavage of the N-O bond. However if this were the case we might expect formation of some 3-benzyloxy-4,6-diphenyl-2pyridone from the decomposition of the 1-benzyloxy-isomer; at the present time we have no evidence for this product in the reaction mixture. We believe a detailed mechanistic study of the whole series of rearrangements is warranted and indeed this is being considered by Dr. I. Shinkai.

Irradiation at 253.7 nm refers to the use of low-pressure mercury lamps (No. RPR-2537A), with 84% of emission at 253.7 nm. Irradiation at 350 nm refers to the use of black-light lamps (No. RPR-3500A) with 90% of emission in the 310 to 410 nm region (maximum at 350 nm). All photolyses were carried out in purified MeCN. Prior to irradiation, the reaction mixture was thoroughly flushed with a stream of dry nitrogen (for at least 1 h) to remove oxygen. During photolysis the reaction solution was kept free of oxygen and dry by the passage of a slow stream of dry nitrogen and the connection of a drying tube.

The following compounds were prepared according to literature procedures: the sodium salt (10) of 1-hydroxy-4,6-diphenyl-2-pyridone (pyridona), m.p. 299—301 °C (lit.,³ 299—301 °C); phenylacetyl chloride, b.p. 47—51 °C at 1.0 mmHg (lit.,³⁰ 104—105 °C at 24 mmHg); o-toluoyl chloride, b.p. 73.5—74.5 °C at 5 mmHg (lit.,³¹ 213 °C at 760 mmHg); p-toluoyl chloride, b.p. 42—45 °C at 0.4 mmHg (lit.,³² 102 °C at 15 mmHg); N-phenylbenzimidoyl chloride (23a), b.p. 104—106 °C at 0.25 mmHg (lit.,¹¹ b.p. 174—176 °C at 12 mmHg); N-(p-tolyl)benzimidoyl chloride (23c), b.p. 138—140 °C at 0.1 mmHg (lit.,¹² b.p. 104 °C at 0.005 mmHg).

Pyrolysis of 1-Phenethyloxy-4,6-diphenyl-2-pyridone (1).—1-Phenethyloxy-4,6-diphenyl-2-pyridone ³ (0.96 g, 2.6 mmol) was heated in an evacuated Claisen distillation flask at 180—220 °C and 1.0 mmHg for 4 h. The cooled receiver contained phenylacetaldehyde (0.15 g, 49%). 3-Benzyl-4,6-diphenyl-2-pyridone (3) (0.023 g, 26%) was separated from 4,6-diphenyl-2-pyridone ^{3b} by extracting the residue with hot Me₂CO (20 ml); the former was collected by filtration. Crystallisation from cyclohexane gave prisms, m.p. 165—167 °C (Found: C, 85.1; H, 6.0; N, 4.4 C₂₄-H₁₉NO requires C, 85.4; H, 5.7; N, 4.2%); v_{max} (CHBr₃) 2 800—3 100 (N-H) and 1 630 cm⁻¹ (C=O); $\delta[(CD_3)_2SO]$ 3.7 (2 H, s), 6.4 (1 H, s), 6.8—7.7 (15 H, m), and 12.0 (1 H, br s); m/e 337 (M^+).

Photolysis of 1-Phenethyloxy-4,6-diphenyl-2-pyridone (1). A solution of the 1-phenethoxypyridone (1.10 g, 3.0 mmol) in EtOAc (100 ml) was deoxygenated and irradiated for 4.5 h in a Pyrex photolysis reactor equipped with a Hanovia 100-W medium-pressure mercury lamp. The solvent was evaporated off and the residue steam-distilled. The residue of the steam distillation was eluted on a silica-gel column (Kieselgel 60PF254). The silica was first heated under reflux in MeOH to remove the indicator. The eluant was first Et₂O (100 ml), then Et₂O-EtOAc (100 ml, 1:1), then EtOAc (200 ml). The third fraction to be eluted (under pressure) was the above-mentioned 3-benzyl-4,6-diphenyl-2-pyridone (0.14 g, 14%), which crystallised from cyclohexane as prisms, m.p. 165—167 °C.

2-Chloro-4,6-diphenylpyridine (6).—4,6-Diphenyl-2-pyridone (2.00 g, 8.1 mmol) and POCl₃ (5 ml) were heated in a Carius tube at 200 °C for 4.5 h. The black solution was poured onto ice-water (500 g, 100 ml) maintained basic with NH₄OH (to phenolphthalein). The precipitate was crystallised from EtOH to give the chloropyridine (6) (1.82 g, 84%) as pale tan prisms, m.p. 63—65 °C (Found C, 76.5; H, 4.6; N, 5.2; Cl, 13.4. $C_{17}H_{12}ClN$ requires C, 76.8; H, 4.5; H, 5.3; Cl, 13.3%); v_{max} . 1 600 cm⁻¹ (C=N, C=C); $\delta(CDCl_3)$ 7.5—8.4 (11 H, m) and 8.4 (1 H, s).

2,4-Diphenylpyridine. 2-Chloro-4,6-diphenylpyridine (0.53 g, 2 mmol), KOH (0.16 g), Pd-C (5%, 0.20 g), and MeOH (50 ml) were stirred under H_2 at 18 °C and 760 mmHg for 18 h. The mixture was filtered and evaporated.

The residue was dissolved in $\rm H_2O$ (100 ml) and extracted with CHCl₃ (3 \times 75 ml). The combined dried (MgSO₄) extracts were evaporated, leaving 2,4-diphenylpyridine as an intractible oil; $\nu_{max.}$ (liquid film) 1 600 cm⁻¹ (C=N, C=C); $\delta(\rm CDCl_3)$ 7.4—8.4 (12 H, m) and 8.7 (1 H, d, J 10 Hz). The oil was converted into the picrate (0.74 g, 80%) which crystallised from aqueous EtOH (95%), m.p. 190.5—192.5 °C [lit.,^{33} m.p. 187 °C].

3-Benzyl-2-chloro-4,6-diphenylpyridine (8).—3-Benzyl-4,6diphenyl pyridone (9) (0.66 g, 1.95 mmol) and POCl₃ (2.0 ml) were heated at 200 °C for 4 h in a Carius tube. The cooled POCl₃ solution was poured into ice-water (100 g, 25 ml) maintained basic with NH₄OH (to phenolphthalein). The aqueous suspension was extracted with CHCl₃ (3 × 50 ml) and the combined organic layers were washed with H₂O (3 × 50 ml). After drying (MgSO₄), the solvent was evaporated off, leaving the 3-benzyl-2-chloro-4,6-diphenylpyridine (8) which crystallised from CHCl₃-MeOH as prisms (0.67 g, 96%), m.p. 148.5—150.5 °C (Found: C, 80.6; H, 5.2; N, 3.9; Cl, 10.1. C₂₄H₁₈ClN, requires C. 81.0; H, 5.1; N, 3.9; Cl, 10.0%); $v_{max.}$ (CHBr₃) 1 600 (C=N), 1 590 (C=C), and 1 150 cm⁻¹ (CCl); δ (CDCl₃) 4.15 (2 H, s), 6.9—7.5 (13 H, m), 7.6 (1 H, s), and 8.0—8.2 (2 H, m).

5-Benzyl-2,4-diphenylpyridine (9).-3-Benzyl-2-chloro-4,-6-diphenylpyridine (8) (0.46 g, 1.3 mmol), KOH (0.20 g), Pd-C (5%, 0.40 g), and EtOH (50 ml) were shaken under H₂ at 15 °C and 2 atm for 18 h. The mixture was filtered, the filtrate was evaporated, and the residue was dissolved in H₂O (60 ml) and extracted with CHCl₃ (3 \times 60 ml). The extracts were combined, dried (MgSO₄), and evaporated, leaving a clear oil (0.41 g, 98%) which did not crystallise; $\nu_{max.}$ (CHBr₃) 1 600 (C=N) and 1 590 cm⁻¹ (C=C); δ (CDCl₃) 4.0 (2 H, s), 6.9–7.6 (13 H, m), 7.8 (1 H, s), 8.1-8.3 (2 H, m), and 8.7 (1 H, s). The crude pyridine (9) was converted into the hydrochloride salt, which crystallised from EtOH as needles, m.p. 196-198 °C (Found: C, 80.3; H, 5.7; N, 3.9; Cl, 10.0. C₂₄H₂₀ClN requires C, 80.5; H, 5.6; N, 3.9; Cl, 9.9%); $\nu_{max.}$ (CHBr₃) 2 300 (broad, NH) and 1 600 cm⁻¹ (C=C, C=N); $\delta[(CD_3)_2SO]$ 4.2 (2 H, s), 6.9-7.7 (13 H, m), 8.2 (1 H, s), 8.3-8.4 (2 H, m), and 8.8 (1 H, s).

Pyrolysis of 1-(But-3-envloxy)-4,6-diphenyl-2-pyridone. 1-(But-3-envloxy)-4,6-diphenyl-2-pyridone³ (0.50 g, 1.6 mmol) was distilled at 200-230 °C and 15 mmHg over 3 h. After this period, the cooled receiver (-78 °C) held an oil (0.055 g, 50%) which was shown to be crotonaldehyde; v_{max} (liquid film) 1 700 cm⁻¹ (C=O); δ (CCl₄) 2.0 (3 H, d, $\int 11$ Hz), 6.1 (1 H, m), 6.7 (1 H, m), and 9.5 (1 H, d, $\int 11$ Hz). The pyrolysis residue was eluted on an alumina column (EtOAc), and the second fraction was 3-allyl-4,6diphenyl-2-pyridone (0.15 g, 33%), which, after sublimation (185 °C at 15 mmHg) was obtained as needles, m.p. 187-189 °C (Found: N, 4.6. $C_{20}H_{17}NO$ requires B, 4.9%); (CHBr₃) 2 800-3 200 (NH) and 1 610 cm⁻¹ (C=O); ν_{max.} δ(CDCl₃) 3.2 (2 H, d), 4.9 (2 H, m), 5.9 (1 H, m), 6.3 (1 H, s), 7.3 (10 H, m), and 12.5 (1 H, br s).

Pyrolysis of 1-Octyloxy-4,6-diphenyl-2-pyridone (11).--1-Octyloxy-2,4-diphenyl-2-pyridone (0.75 g, 2 mmol) was slowly distilled at 200-240 °C and 15 mmHg. After 4 h, the cooled receiver (-78 °C) held a mixture (0.13 g) of octanal (37%) and octan-1-ol (13%). The residue (0.60 g) was separated by preparative t.l.c. (silica; CHCl₃, 4 developments). The third band from the baseline was 3octyloxy-4,6-diphenyl-2-pyridone (13) (0.04 g, 5%) which crystallised from $CHCl_3$ -light petroleum (b.p. 60–80 °C) as needles, m.p. 133–136 °C (Found: C, 79.6; H, 7.6; N, 3.6. $C_{25}H_{29}NO_2$ requires C, 80.0; H, 7.8; N, 3.7%); ν_{max} . (CHBr₃) 2 700–3 100 (NH), 1 600–1 650 (C=O), and 1 040 cm⁻¹ (C=O); δ (CDCl₃) 1.6 (15 H, m), 4.0 (3 H, t, J 8 Hz), 6.5 (1 H, s), 7.3–7.9 (10 H, m), and 12.7 (1 H, br s).

Photolysis of 1-Octyloxy-4,6-diphenyl-2-pyridone (11).—A solution of 1-octyloxy-4,6-diphenyl-2-pyridone (11) (1.13 g, 3.0 mmol) in dry, deoxygenated EtOAc (100 ml) was irradiated for 4 h in a pyrex photolysis reactor equipped with a Hanovia 100-W medium-pressure mercury lamp. The solvent was evaporated off and the residue steam-distilled, giving octanal (47%) and octanol (23%). On preparative t.l.c. (silica; toluene-EtOAc, 2:1) the distillation residue gave the above-mentioned 3-octyloxypyridone (0.02 g, 2%).

1-(5-Nitro-2-pyridyloxy)-4,6-diphenyl-2-pyridone (12). Sodium 1-oxido-4,6-diphenyl-2-pyridone (10) (0.95 g, 3.3 mmol), 2-chloro-5-nitropyridine (0.48 g, 3.0 mmol), and absolute EtOH (40 ml) were heated under reflux for 7 h. The solution was cooled and filtered to give the *pyridyloxypyridone* (12) (1.11 g, 96%) as prisms which crystallised from EtOH, m.p. 190–192 °C (Found: C, 68.4; H, 4.1; N, 10.7. $C_{22}H_{15}N_3O_4$ requires C, 68.6; H, 3.9; N, 10.9%); ν_{max} (CHBr₃) 1 670 (C=O) and 1 360 cm⁻¹ (NO₂); δ (CDCl₃) 6.5 (1 H, d, J 2 Hz), 6.9 (1 H, d, J 2 Hz), 7.0 (1 H, d, J 9 Hz), 7.3–7.7 (10 H, m), 8.4 (1 H, dd, J 9 Hz, J 3 Hz), and 9.0 (1 H, d, J 3 Hz).

Photolysis of 1-(5-Nitro-2-pyridyloxy)-4,6-diphenyl-2pyridone (12).—A dry, deoxygenated solution of 1-(5nitro-2-pyridyloxy)-4,6-diphenyl-2-pyridone (0.93 g, 2.4 mmol) in benzene (100 ml) was irradiated for 3 h in a Pyrex photolysis reactor equipped with a Hanovia 100-W mediumpressure mercury lamp. The benzene was evaporated off and the residue taken up in acetone. Filtration gave a dark brown microcrystalline solid, 3-(5-nitro-2-pyridyloxy)-4,6-diphenyl-2-pyridone (14) (0.46 g, 50%) which resisted purification; v_{max} . (CHBr₃) 3 200—2 800 (N-H), 1 670— 1 620 (C=O), 1 350 (NO₂), and 1 260 cm⁻¹ (C-O); δ (CDCl₃), 6.5 (1 H, d, J 10 Hz), 6.7 (1 H, s), 7.4 (10 H, m), 8.1 (1 H, dd, J 10, 2 Hz), and 8.4 (1 H, d, J 2 Hz).

1-Benzoyloxy-4,6-diphenyl-2-pyridone (15e).—Benzoyl chloride (1.12 g, 0.008 mol), the salt (10) (1.14 g, 0.004 mol), and MeCN (25 ml) were stirred for 2.5 h at 20 °C. Solid was filtered off and the filtrate was reduced to 5 ml and again filtered. The combined precipitates were recrystallised from EtOAc to give colourless needles of 1-benzoyloxy-4,6-diphenyl-2-pyridone (15e) (1.35 g, 92%), m.p. 215—216 °C (Found: C, 78.2; H, 4.6; N, 4.0. C₂₄-H₁₇NO₃ requires C, 78.5; H, 4.7; N, 3.8%); ν_{max} . (Nujol) 1 780 (O-C=O) and 1 670 cm⁻¹ (N-C=O); λ_{max} . (EtOH) 341 and 244 nm; δ (CDCl₃) 6.44 (1 H, d, J 3 Hz), 6.90 (1 H, d, J 3 Hz), 7.2—7.7 (13 H, m), and 7.95 (2 H, d of d, J 9 and 1.5 Hz).

The following were prepared similarly: 1-acetoxy-4,6-diphenyl-2-pyridone (15a) (76%), needles from dry EtOAc-dry light petroleum (b.p. 60—80 °C), m.p. 123— 124 °C (Found: C, 74.6; H, 5.0; N, 4.6. $C_{19}H_{15}NO_3$ requires C, 74.7; H, 5.0; N, 4.6%); v_{max} . (Nujol) 1 815 (O-C=O) and 1 670 cm⁻¹ (N-C=O); δ (CDCl₃) 2.10 (3 H, s, Me), 6.44 (1 H, d, J 2.5 Hz), 6.82 (1 H, d, J 2.5 Hz), and 7.50 (10 H, s, 2Ph); 4,6-diphenyl-1-phenylacetoxy-2pyridone (15b) (68%) not purified; v_{max} . (Nujol) 1 810 (O-C=O) and 1 670 cm⁻¹ (N-C=O); δ (CDCl₃) 3.72 (2 H, s, CH₂), 6.40 (1 H, d, J 2.5 Hz), 6.70 (1 H, d, J 2.5 Hz), and 7.0—7.7 (17 H, m); 4,6-diphenyl-1-(o-toluoyloxy)-2-pyridone (15c) (66%), prisms from dry EtOAc-dry light petroleum (b.p. 60—80 °C), m.p. 133.5—135 °C (Found: C, 78.7; H, 5.2; N, 3.6. $C_{25}H_{19}NO_3$ requires C, 78.7; H, 5.0; N, 3.7%); v_{max} . (Nujol) 1 780 (O-C=O) and 1 660 cm⁻¹ (N-C=O); δ (CDCl₃) 2.44 (3 H, s, Me), 6.50 (1 H, d, J 2.5 Hz), 6.98 (1 H, d, J 2.5 Hz), and 7.5 (14 H, m); 4,6-diphenyl-1-(ptoluoyloxy)-2-pyridone (15d) (95%), needles from dry EtOAc, m.p. 213—214 °C (Found: C, 78.8; H, 5.0; N, 3.7. $C_{25}H_{19}NO_3$ requires C, 78.7; H, 5.0; N, 3.7%); v_{max} (Nujol) 1 780 (O-C=O) and 1 660 (N-C=O), δ (CDCl₃) 2.45 (3 H, s, Me), 6.56 (1 H, d, J 2.5 Hz), 6.96 (1 H, d, J 2.5 Hz), 7.23 (2 H, d, J 8 Hz), 7.5 (10 H, m), and 7.90 (2 H, d, J 8 Hz).

Photolysis of 1-Benzoyloxy-4,6-diphenyl-2-pyridone (15e).-1-Benzoyloxy-4,6-diphenyl-2-pyridone (1.14 g, 3.1 mmol) in dry MeCN (250 ml) was degassed and irradiated at 350 nm for 36 h. The solution was reduced to 25 ml, cooled overnight at 0 °C, and filtered. The precipitate crystallised from dry MeCN to give 5-benzoyloxy-4,6diphenyl-2-pyridone (16) (140 mg, 12%) as needles, m.p. 249–252 °C (Found: C, 78.1; H, 4.5; N, 3.8. $C_{24}H_{17}NO_3$ requires C, 78.5; H, 4.7; N, 3.8%); ν_{max} (Nujol) $3\ 000$ — 2 500 (NH), 1 740 (O–C=O), and 1 650 cm⁻¹ (N–C=O); δ(CDCl₃) 6.6 (1 H, s, 3-H), 7.2-7.7 (13 H, m), and 7.8 (2 H, d, J 9 Hz); m/e (70 eV) 368 (8.2%), 367 (26), 262 (4.1), 247 (1.5), 234 (2.4), 206 (3.6), 128 (1.6), 122 (33), 105 (100), 83 (11), 77 (60), 66 (9.3), 57 (5.7), 51 (10), and 45 (9.8); λ_{\max} (CHCl₃) 335 nm.

The filtrate was evaporated and the residue separated by preparative t.l.c. (silica), eluting with EtOAc, to give 4,6diphenyl-2-pyridone [4th fraction, $R_{\rm F}$ 0.30 (EtOAc, silica)]; recrystallised from Me₂CO (80 mg, 10%), m.p. 212—215 °C, mixed m.p. 210—213 °C, m.p. (authentic sample) 210— 213 °C (lit.,³⁴ 208 °C), identical (i.r. and ¹H n.m.r. spectra) with an authentic sample, benzoic acid [2nd fraction, $R_{\rm F}$ 0.80 (EtOAc, silica)]; isolated by dissolving in CHCl₃, extraction with aqueous Na₂CO₃, acidification, re-extraction with CHCl₃, drying, and removal of solvent (9 mg, 2%), m.p. 120—122 °C (lit.,³⁵ m.p. 122 °C), identical (i.r. spectrum) with an authentic sample].

Hydrolysis of 5-Benzoyloxy-4,6-diphenyl-2-pyridone (16).-5-Benzoyloxy-4,6-diphenyl-2-pyridone (16) (60 mg, 16 mmol) was refluxed in aqueous NaOH (20%, 5 g) with stirring for 3.5 h. On cooling, water (45 ml) was added and the mixture acidified with concentrated HCl and then basified to pH 8.5 with Na₂CO₃. The precipitate was washed with water (10 ml), and recrystallised from aqueous HOAc to give 5-hydroxy-4,6-diphenyl-2-pyridone (17) (31 mg, 71%) as needles, m.p. 273-276 °C (Found: C, 77.4; H, 4.9; N, 5.2. C₁₇H₁₃NO₂ requires C, 77.5; H, 5.0; N, 5.3%); (Nujol) 3 500–2 500 (OH) and 1 640 cm⁻¹ (N-C=O); $\delta(CF_3CO_2H)$ 7.25 (1 H, s, 3-H) and 7.65 (10 H, m); m/e264 (19%), 263 (100), 262 (31), 247 (8.1), 244 (4.1), 235 (9.0), 234 (4.3), 206 (18), 191 (2.1), 130 (5.3), 105 (17), 104 (17), 102 (6.9), and 77 (12); λ_{max} (CHCl₃) 333 nm $(11 \ 300).$

The aqueous filtrate was acidified with concentrated HCl and extracted with $CHCl_3$ (3×20 ml). The $CHCl_3$ extracts were combined, dried (MgSO₄), and evaporated. The residue was recrystallised from light petroleum (b.p. 60—80 °C) to give benzoic acid, (19 mg, 68%), m.p. 121—122 °C, mixed m.p. 121—122 °C, identical (i.r. spectrum) with an authentic sample.

Photolysis of 4,6-Diphenyl-1-(p-toluoyloxy)-2-pyridone

(15d).—4,6-Diphenyl-1-(p-toluoyloxy)-2-pyridone (3.0 g, 7.9 mmol) in dry MeCN (500 ml) was degassed and irradiated at 350 nm for 38 h. The photolysis mixture on filtration yielded a precipitate (a). The filtrate was reduced to 30 ml and filtered again to afford a further precipitate (b). $CHCl_3$ (25 ml) was added to the precipitate (a) and the suspension was boiled for 5 min, cooled and filtered to give a precipitate (c) (405 mg). The filtrate was added to the precipitate (b) and the suspension boiled for 5 min, cooled, and filtered to give another precipitate (d) (52 mg). The filtrate was eluted down an alumina column with CHCl₃-EtOH (9:1 v/v). The first fraction (35 mg), identical to precipitates (c) and (d), was 4,6-diphenyl-3-(p-toluoyloxy)-2pyridone (19) (total yield 492 mg, 16%), which crystallised from dry MeCN as needles, m.p. 290-295 °C (Found: C, 78.6; H, 5.0; N, 3.7. C₂₅H₁₉NO₃ requires C, 78.7; H, 5.0; N, 3.7%); ν_{max} (Nujol) 3 100–2 500 (NH), 1 740 (O-C=O), and 1.660 cm^{-1} (N-C=O); $\delta(\text{CF}_3\text{CO}_2\text{H}) 2.4$ (3 H, s, Me), 7.2-7.9 (13 H, m) and 8.0 (2 H, half of an ABq, J 8 Hz); $\delta(CDCl_3)$ 2.4 (3 H, s, Me), 6.4 (1 H, s, 5-H), and 7.0–7.4 (14 H, m); λ_{max} (EtOH) 333 nm (ε 14 000); $(CHCl_3)$ 335 nm (15 800); m/e (70 e.v.) 381 (12%), 263 (6.0), 262 (4.3), 243 (3.1), 119 (100), 104 (8.9), 91 (25), 77 (4.7), and 65 (5.9). The second fraction from the column crystallised from dry MeCN to give 4,6-diphenyl-5-(p-toluoyloxy)-2-pyridone (18) (295 mg, 10%) as needles, m.p. 246-252 °C (Found: C, 79.1; H, 5.0; N, 3.7. C25- $\begin{array}{c} H_{19}NO_3 \ requires \ C, \ 78.7; \ H, \ 5.0; \ N, \ 3.7\%); \ \nu_{max.} \ (Nujol) \\ 3 \ 100-2 \ 500 \ (NH), \ 1 \ 740 \ (O-C=O), \ and \ 1 \ 650 \ cm^{-1} \ (N-C=O); \end{array}$ δ(CDCl₃) 2.3 (3 H, s, Me), 6.6 (1 H, s, 3-H), 7.1 (2 H, half of an ABq, J 8 Hz), 7.1–7.7 (10 H, m), and 7.7 (2 H, half of an ABq, y 8 Hz); λ_{max} (EtOH) 327 nm; m/e (70 eV) 381 (22%), 262 (4.3), 247 (4.8), 206 (4.1), 119 (100), 104 (15), 102 (9.6), 91 (27), 77 (15), and 65 (13).

Filtrate (b) was evaporated and the residue separated on an alumina column eluting successively with $CHCl_3$ -EtOAc (4:1 v/v), EtOAc, EtOAc-MeOH, and MeOH. The second fraction $[R_F 0.80$ (EtOAc, silica), 0.55 (EtOAc-CHCl₃, 4:1 v/v; silica)] was *p*-toluic acid (128 mg, 12%), m.p. 179—181 °C (lit.,³⁶ m.p. 181 °C), identical (i.r. spectrum and t.l.c.) with an authentic sample. Also collected was 4,6-diphenyl-2-pyridone (680 mg, 35%) [$R_F 0.30$ (EtOAc, silica); m.p. 209—213 °C (from Me₂CO) (lit.,³⁴ m.p. 208 °C); identical (i.r. spectrum and t.l.c.) with an authentic sample], and several minor unidentified products.

Hydrolysis of 4,6-Diphenyl-5-(p-toluoyloxy)-2-pyridone (18).--4,6-Diphenyl-5-p-toluoyloxy-2-pyridone (18) (800 mg, 2.1 mmol) was refluxed with aqueous NaOH (20%, 20 g) for 75 min with stirring. On cooling, water (40 ml) was added and the mixture was acidified with concentrated HCl and then re-basified to pH 8.5 with Na₂CO₃. The precipitate was filtered off, washed with water (25 ml), and recrystallised from aqueous HOAc to give 5-hydroxy-4,6-diphenyl-2-pyridone (354 mg, 64%) as needles, identical to compound (17) obtained from hydrolysis of 5-benzoyloxy-4,6-diphenyl-2-pyridone (16). The aqueous filtrate was acidified with concentrated HCl. The precipitate was filtered off, washed with water (10 ml), and recrystallised from water to give p-toluic acid (189 mg, 66%), m.p. 180-182 °C, identical (i.r. spectrum, m.p.) with an authentic sample.

4,6-Diphenyl-3-(p-toluoyloxy)-2-pyridone (19) was hydrolysed similarly to give p-toluic acid (69%), m.p. 180— 182 °C, and 3-hydroxy-4,6-diphenyl-2-pyridone (20) (70%) as needles (from Me₂CO), m.p. 204—205 °C (Found: C, 77.3; H, 4.9; N, 5.3. $C_{17}H_{13}NO_2$ requires C, 77.5; H, 5.0; N, 5.3%); ν_{max} (Nujol) 3 350, 3 060, and 1 630 cm⁻¹ (N-C=O); δ (CDCl₃) 6.6 (1 H, s, 5-H), 7.2—7.8 (10 H, m), and 12.5 (1 H, disappears on shaking with D₂O); λ_{max} (CHCl₃) 335 nm (ε 11 300); m/e (70 eV) 263 (100%), 262 (66), 244 (5.9), 234 (2.9), 217 (7.8), 216 (8.5), 206 (5.3), 191 (3.0), 189 (6.3), 128 (1.9), 104 (10), and 77 (8.3).

2-Chloro-4,6-diphenyl-3-(p-toluoyloxy)pyridine (21).—4,6-Diphenyl-3-(p-toluoyloxy)-2-pyridone (19) (570 mg, 1.5 mmol) in PCl₃ (15 ml) was heated in a Carius tube at 200 °C for 2 h. The resultant yellow liquid was poured into ice-water (600 g: 150 ml) containing sufficient aqueous NH₄OH to maintain the solution basic (to phenolphthalein). The precipitate was filtered off, washed with water (250 ml), and dried to give the crude 2-chloro-3-(p-toluoyloxy)pyridine (21) (164 mg, 28%). Recrystallisation from aqueous EtOH (95%) afforded needles, m.p. 198—199 °C (Found: C, 74.7; H, 4.4; N, 3.4. $C_{25}H_{18}CINO_2$ requires C, 75.1; H, 4.5; N, 3.5%); v_{max} (Nujol) 1 745 cm⁻¹ (C=O); δ (CDCl₃) 2.4 (3 H, s, Me) and 7.1—8.2 (15 H, m).

The aqueous filtrate and washings were extracted with $CHCl_3$ (2 × 150 ml). The $CHCl_3$ extracts were combined, washed with water (2 × 50 ml), and dried (MgSO₄), and the solvent removed. The solid residue was crystallised from $CHCl_3$ to give *p*-toluamide (92 mg, 47%) as needles, m.p. 159—160 °C (lit.³⁷ m.p. 165 °C).

2,4-Diphenylpyridin-5-ol (22).-Pd-C (5%, 100 mg), 2-chloro-4,6-diphenyl-3-(p-toluoyloxy)pyridine (98 mg. 0.245 mmol), KOH (35 mg), and MeOH (50 ml) were stirred under H₂ at 20 °C and 760 mmHg for 69 h. The mixture was filtered and the filtrate evaporated. Water (50 ml) was added to the residue and this solution was extracted with $CHCl_3$ (3 \times 25 ml). The combined dried (MgSO₄) extracts were evaporated to leave the 3-hydroxypyridine (22) (54 mg, 89%) which was recrystallised from CHCl_a as needles (38 mg, 63%), m.p. 184-185 °C with some sublimation at ca. 180 °C (Found: C, 82.2; H, 5.1; N, 5.6. $C_{17}H_{13}NO$ requires C, 82.6; H, 5.3; N, 5.7%); v_{m} (Nujol) 2800-2500 cm⁻¹ (broad, OH); $\delta[(CD_3)_2SO]$ 7.2-7.6 (6 H, m, m-H, m'-H, p-H, p'-H), 7.6-7.9 (3 H, m, o'-H, 3-H), 7.9-8.2 (2 H, m, o-H), and 8.5 (s, 6-H).

α-(p-Anisidino)-o-tolualdehyde (31).—o-Toluoyl chloride (7.87 g, 0.05 mol) was shaken with p-anisidine (5.53 g, 0.045 mol) in aqueous NaOH (10%, 37 ml) for 15 min. After dilution with water (50 ml), the precipitate was filtered off, washed with water (50 ml), and recrystallised from EtOH to give the tolualdehyde (31) as needles (5.8 g, 55%), m.p. 143—144.5 °C (Found: C, 74.9; H, 6.3; N, 5.8. $C_{15}H_{15}NO_2$ requires C, 74.7; H, 6.3; N, 5.8%); ν_{max} . (Nujol) 3 290 (NH) and 1 650 cm⁻¹ (C=O); δ (CDCl₃) 2.44 (3 H, s), 3.78 (3 H, s), 6.93 (2 H, d, J 9 Hz), 7.3 (4 H, m), and 7.76 (2 H, d, J 9 Hz).

N-(p-Methoxyphenyl)-o-toluimidoyl Chloride (23b).—Thionyl chloride (7 g) and α-(p-anisidino)-o-tolualdehyde (8.7 g, 0.036 mol) were refluxed for 1.5 h. The excess of thionyl chloride was removed and the product was distilled to give the *imidoyl chloride* (23b) (8.1 g, 86%) as a pale yellow oil, b.p. 152—155 °C at 0.05 mmHg, which solidified to needles (Found: C, 69.5; H, 5.8; N, 5.3. C₁₅H₁₄ClNO requires C, 69.4; H, 5.4; N, 5.4%); $\nu_{max.}$ (Nujol) 1 645 (C=N) cm⁻¹; δ (CDCl₃) 2.50 (3 H, s), 3.82 (3 H, s), 7.05 (2 H, d, J 4 Hz), 7.3 (4 H, m), and 7.70 (2 H, d, J 4 Hz).

4,6-Diphenyl-1-(N-phenylbenzimidoyloxy)-2-pyridone (24a).—N-Phenylbenzimidoyl chloride (23a) (0.22 g, 1 mmol), sodium 1-oxido-4,6-diphenyl-2-pyridone (10) (0.29 g, 1 mmol), and dry MeCN (10 ml) were stirred for 6 h at 20 °C with exclusion of light and water vapour. The reaction mixture was filtered through Kieselguhr. The solvent was removed from the filtrate and the residue recrystallised from dry Et₂O to give the 1-benzimidoyl-oxypyridone (24a) (0.39 g, 88%) as needles, m.p. 147—148 °C (Found: C, 81.0; H, 5.2; N, 6.4. $C_{30}H_{22}N_2O_2$ requires C, 81.4; H, 5.0; N, 6.3%); ν_{max} . (Nujol) 1 660 cm⁻¹ (C=O); δ (CDCl₃) 6.44 (1 H, d, J 2.5 Hz), 6.56 (2 H, d, J 8 Hz), 6.94 (1 H, d, J 2.5 Hz), and 6.9—7.8 (20 H, m).

The following were prepared similarly: 1-[N-(p-methoxyphenyl)-o-toluimidoyloxy)-4,6-diphenyl-2-pyridone (24b) (93%) as needles from dry Et₂O, m.p. 143—145 °C (Found: C, 78.6; H, 5.4; N, 5.7. $C_{32}H_{26}N_2O_3$ requires C, 79.0; H, 5.4; N, 5.8%); v_{max} . (Nujol) 1 660 cm⁻¹ (C=O); δ (CDCl₃) 1.8 (3 H, s), 3.64 (3 H, s), 6.5 (4 H, s), 6.54 (1 H, d, J 2.5 Hz), 6.95 (1 H, d, J 2.5 Hz), and 7.0—7.8 (14 H, m); 4,6diphenyl-1-[N-(p-toluoyl)benzimidoyloxy]-2-pyridone (24c) (95%) as needles (from dry EtOAc), m.p. 147—149 °C (Found: C, 81.3; H, 5.2; N, 6.1. $C_{31}H_{24}N_2O_2$ requires C, 81.6; H, 5.3; N, 6.1%); v_{max} . (Nujol) 1 660 cm⁻¹ (C=O); δ (CDCl₃) 2.19 (3 H, s, Me), 6.45 (1 H, d, J 2.5 Hz), 6.5 (2 H, d, J 8 Hz), 6.95 (2 H, d, J 8 Hz), 7.0 (1 H, d, J 2 Hz), and 7.1—7.8 (15 H, m).

Photolysis of 4,6-Diphenyl-1-(N-phenylbenzimidoyloxy)-2pyridone (24a).—The benzimidoyloxypyridone (24a) (7.78 g, 17.6 mmol) in dry MeCN (500 ml) was degassed and irradiated at 350 nm for 40 h. The precipitate was filtered off and recrystallised from CF₃CO₂H to give needles of 4,4'dibenzamidobiphenyl (25) (60 mg, 1.7%), m.p. >335 °C [lit.,¹³ m.p. 352 °C] (Found: C, 79.6; H, 5.2; N, 7.1. C₂₆H₂₀N₂O₂ requires C, 79.6; H, 5.1; N, 7.1%); $\nu_{\text{max.}}$ (Nujol) 3 240 (NH) and 1 645 cm⁻¹ (N-C=O); δ (CDCl₃) 7.7 (18 H, m), and 9.4 (2 H, s, NH); m/e (70 eV) 392 (63%), 391 (2.1), 289 (5.8), 287 (3.7), 105 (100), 77 (63), and 44 (6.6).

Removal of solvent from the filtrate, addition of CHCl₃ (30 ml) to the residue and cooling overnight at 0 °C gave crystalline 5-benzanilido-4,6-diphenyl-2-pyridone (26) (980 mg, 12.5%), which recrystallised from CHCl₃ as needles, m.p. 306-308 °C (Found: N, 6.1. $C_{30}H_{22}N_2O_2$ requires N, 6.3%); ν_{max} (Nujol) 1 645 cm⁻¹; δ (CF₃CO₂H) 7.0-8.0 (m); λ_{max} (CHCl₃) 293 (22 300) and 340 nm (shoulder).

The filtrate was eluted from an alumina column successively with CHCl_a and CHCl_a-MeOH (19:1 v/v). The 3rd fraction collected was benzanilide (140 mg, 4%), m.p. 160-163 °C (lit. 38 m.p. 162 °C), identical with authentic sample (i.r. spectrum). The 10th fraction crystallised from dry MeCN to give 3-benzanilido-4,6-diphenyl-2pyridone (28) (362 mg, 4%) as needles, m.p. 330-333 °C (Found: C, 81.2; H, 5.2; N, 6.6. C₃₀H₂₂N₂O₂ requires C, 81.4; H, 5.0; N, 6.3%); $\nu_{max.}$ (Nujol) 1 670 and 1 630 cm⁻¹; δ (CF₃CO₂H) 7.2—8.1 (m); $\lambda_{max.}$ (CHCl₃) 354 nm (19 900); m/e (70 eV) 442 (100%), 441 (98). 423 (5.0), 337 (22), 309 (12), 291 (5.0), 280 (6.2), 265 (5.0), 230 (6.2), 203 (6.7), 178 (5.9), 105 (86), 104 (8.4), and 77 (70). The 12th fraction was a mixture of 4,6-diphenyl-2-pyridone 5-benzanilido-4,6-diphenyl-2-pyridone (26). Reand crystallisation from Me₂CO afforded solely 4,6-diphenyl-2pyridone (540 mg, 12.5%), m.p. 210-214 °C (lit., 34 208 °C), identical (i.r. spectrum) with an authentic sample. Removal of solvent from the mother-liquor and recrystallisation of the residue from CHCl₃ afforded solely the 5benzanilidopyridone (26) (80 mg, 1%; total yield 1.06 g, 13.5%). Removal of solvent from the mother-liquor and

recrystallisation from Me₂CO afforded further 4,6-diphenyl-2-pyridone (270 mg, 6%, total yield 810 mg, 18.5%). Also obtained from the column in moderate quantities were two unidentified products (680 mg and 930 mg).

5-Benzanilido-4,6-diphenyl-2-pyridone Hvdrolvsis of (26).—Aqueous H_2SO_4 (70%, 5 g) was added to the 5benzanilidopyridone (26) (216 mg, 0.49 mmol) and the mixture heated at 140 °C for 4 h with stirring. A small quantity of benzoic acid sublimed from the reaction mixture as needles, m.p. 116-122 °C, identical with an authentic sample (i.r., m.p.). After cooling, the reaction mixture was diluted with water (45 ml) and neutralised to pH 8.5 (Na₂CO₃). The precipitate was filtered off, washed with water (50 ml), dried, and recrystallised from EtOH to give 5-anilino-4,6-diphenyl-2-pyridone (27) (132 mg, 80%) as needles, m.p. 334-337 °C (Found: C, 81.5; H, 5.3; N, 8.1. $C_{23}H_{18}N_2O$ requires C, 81.6; H, 5.4; N, 8.3%); (Nujol) 3 440, 3 330, and 1 640 cm⁻¹; δ (CF₃CO₂H⁻ $CDCl_3$) 7.3 (m); $\lambda_{max.}$ (CHCl₃) 348 nm.

The filtrate from above after removal of the precipitate of (27) and washings were combined, acidified (concentrated HCl), and extracted with CHCl_3 (3 × 30 ml). The extracts were combined and dried (MgSO₄) and the solvent was removed. The solid residue was recrystallised from light petroleum (b.p. 60–80 °C) to give benzoic acid (39 mg, 66%), m.p. 120–123 °C.

Hydrolysis of 3-Benzanilido-4,6-diphenyl-2-pyridone (28).—Aqueous H_2SO_4 (70%, 5 g) and 3-benzanilidopyridone (28) (159 mg) were stirred at 150 °C for 105 min. Benzoic acid partially sublimed from the reaction mixture. Or cooling, the reaction mixture was diluted with water (45 ml), and neutralised to pH 8.5 (Na₂CO₃). A dark green precipitate of crude 3-anilino-4,6-diphenyl-2-pyridone (29) (116 mg, 95%) was filtered off, washed with water (50 ml), and dried; λ_{max} . (CHCl₃) 355 nm. The filtrate and washings were combined, acidified (concentrated HCl), and extracted with CHCl₃ (3 × 30 ml) to give benzoic acid (36 mg, 81%).

Photolysis of 1-N-(p-Methoxyphenyl)-o-toluimidoyloxy]-4,6-diphenyl-2-pyridone (24b).-The 1-toluimidoyloxypyridone (7.06 g, 14.5 mmol) in dry MeCN (250 ml) was degassed and irradiated for 48 h at 350 nm. The precipitate was filtered off. Solvent was evaporated off from the filtrate and the residue in EtOAc (25 ml) was cooled overnight. Solid was filtered off, combined with the first precipitate, and recrystallised from dry MeCN to give 4,6-diphenyl-5-[No-toluoyl)-p-anisidino]-2-pyridone (30) (820 mg, 11.5%) as needles, m.p. 296-300 °C (Found: C, 79.0; H, 5.5; N, $C_{32}H_{26}N_2O_3$ requires C, 79.0; H, 5.4; N, 5.8%); 5.8. (Nujol) 1 650 cm⁻¹; δ(CDCl_a) 2.3 (3 H, s, Me), 2.5 (3 H, s, OMe), 6.0 (4 H, AB q, J 9 Hz), 6.5 (1 H, d, 3-H), 7.0 (2 H, d, J 4 Hz), and 7.4 (12 H, m); $\lambda_{\text{max.}}$ (CHCl₃) 327 nm (ε 7 700); m/e (70 eV) 486 (32%), 367 (16), 224 (16), 119 (100), 104 (6.7), 91 (31), 77 (8.9), 65 (5.3), and 44 (11.6).

Removal of solvent from the EtOAc filtrate left a residue which was eluted on alumina with $CHCl_3$ -EtOAc (4:1 v/v) followed by $CHCl_3$ -MeOH (10:1 v/v), to afford N-(o-toluoyl)-p-anisidine (31) (1.40 g, 40%), m.p. 143—145 °C (from ether), identical with an authentic sample (t.l.c., m.p., 'H n.m.r. and i.r.). A further component, on recrystallisation from tetrahydrofuran (THF), gave 4,6-diphenyl-2pyridone (720 mg, 20%), which after recrystallisation from Me₂CO had m.p. 210—214 °C (lit.,³⁴ m.p. 208 °C), identical with an authentic sample (m.p., i.r., and t.l.c.). The 5toluoylanisidinopyridone (30) (120 mg. 1.7%) precipitated from the THF mother-liquor on standing. The residual mixture was separated by preparative t.l.c. eluting twice with CHCl₃-MeCN (1:1 v/v) to give 4,6-diphenyl-2-pyridone ($R_{\rm F}$ 0.25; 180 mg, 5%; total yield 900 mg, 25%) and the 5-toluoylanisidinopyridone (30) ($R_{\rm F}$ 0.30; 140 mg, 2%; total yield 980 mg, 15%).

Hydrolysis of 4,6-Diphenyl-5-[N-(0-toluoyl)-p-anisidino)-2-pyridone (30).—Compound (30) (100 mg, 0.2 mmol) in glacial HOAc (2 ml) and concentrated HCl (13 ml) was heated (sealed tube) at 150 °C for 2.5 h. The reaction mixture was diluted with water to 70 ml, neutralised to pH 8 (Na₂CO₃), and extracted with CHCl₃ (3 × 30 ml). The extracts were combined and dried (MgSO₄) and the solvent removed. Trituration with Me₂CO and recrystallisation from Me₂CO gave 5-amino-4,6-diphenyl-2-pyridone (32) (18 mg, 33%) as a brown powder, m.p. 228—233 °C; ν_{max} (Nujol) 3 430 (NH₂), 3 300 (NH₂), and 1 655 cm⁻¹ (C=O); δ (CDCl₃) 2.2 (2 H, s, NH₂), 6.5 (1 H, s, 3-H), and 7.5 (10 H, m); m/e (70 eV) 262 (100%), 261 (34), 243 (7.5), 234 (19), 216 (8.9), 206 (8.5), 130 (20), 104 (17), 102 (11), and 77 (17).

Photolysis of 4,6-Diphenyl-1-[N-(p-tolyl)benzimidoyloxy]-2-pyridone (24c).—The 1-benzimidoyloxypyridone (24c) (7.8 g, 17.1 mmol) in dry MeCN (500 ml) was degassed and irradiated for 36 h at 350 nm. After filtration, the filtrate was reduced to 30 ml, cooled overnight at 0 °C, and filtered again. The precipitates were combined and recrystallised from dry MeCN to yield 5-(N-benzoyl-p-toluidino)-4,6diphenyl-2-pyridone (35) (810 mg, 10%) as needles, m.p. 259–261.5 °C (Found: C, 81.2; H, 5.2; N, 6.0. $C_{31}H_{24}$ -N₂O₂ requires C, 81.6; H, 5.3; N, 6.1%); ν_{max} (Nujol) 1 645 cm⁻¹ (C=O); δ (CF₃CO₂H) 2.28 (3 H, s, Me), 6.9 (1 H, s, 3-H), and 7.5 (19 H, m); $\delta(\text{CDCl}_3)$ 2.1 (3 H, s, Me), 6.6 (1 H, s, 3-H), and 7.3 (19 H, m); λ_{max} , (CHCl₃) 332 (9 000); m/e (70 eV) 456 (51%), 381 (24), 352 (7.4), 335 (22), 323 (5.9), 307 (23), 274 (6.9), 105 (100), and 77 (33). On standing, a precipitate formed in the filtrate. This was recrystallised from dry MeCN to yield 3-(N-benzoyl-ptoluidino)-4,6-diphenyl-2-pyridone (33) (270 mg, 3.5%) as needles, m.p. 305-308 °C (Found: C, 81.2; H, 5.2; N, 6.5. $C_{31}H_{24}N_2O_2$ requires C, 81.6; H, 5.3; N, 6.1%); $\nu_{\rm max.}~({\rm Nujol})~1~670~{\rm cm^{-1}}~({\rm C=O});~\lambda_{\rm max.}~({\rm CHCl_3})~351~{\rm nm}~(\varepsilon$ 15 000); $m/e~(70~{\rm eV})~456~(44\%),~439~(35),~379~(15),~351$ (9.0), 335 (94), 293 (5.5), 105 (100), and 77 (44). The solvent was removed from the filtrate and the residue separated on an alumina column eluting successively with CHCl_a, $CHCl_3$ -EtOAc (4:1 v/v) and $CHCl_3$ -MeOH (4:1 v/v). The 3rd fraction was recrystallised from EtOH to afford N-benzyl-p-toluidide (500 mg, 14%), m.p. 159-161 °C (lit.,³⁹ m.p. 156-158 °C), identical with an authentic sample (i.r., t.l.c.). A further fraction was recrystallised from Et₂O to give compound (39) or (40) (460 mg, 17%) as plates, m.p. 158-159 °C (Found: C, 79.4; H, 6.0; N, 4.3. $C_{21}H_{19}NO_2$ requires C, 79.5; H, 6.0; N, 4.4%); v_{max.} (Nujol) 1 660, 1 640, 1 625, 1 600, 1 575, 1 510, 1 345, 855, and 700 cm⁻¹; δ (CDCl₃) 1.6 (3 H, s, Me), 2.2 (3 H, s, Me), 6.2 (2 H, d, J 10 Hz), and 7.8-8.4 (11 H, m); m/e (70 eV) 317 (4.15%), 211 (31), 107 (100), 105 (94), 90 (13), 77 (75), 51 (23), and 39 (15). Subsequent fractions contained more 3-benzoyltoluidinopyridone (33) (total yield 1.25 g, 16%), and a mixture of 4,6-diphenyl-2-pyridone (500 mg, 12%) and more 5-benzoyltoluidinopyridone (35) (total yield 1.40 g, 18%); the ratio was determined from integration of the ¹H n.m.r. signals.

Hydrolysis of 3-(N-Benzoyl-p-toluidino)-4,6-diphenyl-2pyridone (33).—Aqueous H_2SO_4 (70%, 10 g) and the 3benzoyltoluidinopyridone (33) (473 mg, 1.04 mmol) were heated at 140 °C for 1.5 h with stirring. Some benzoic acid sublimed. On cooling, the mixture was diluted with water (40 ml) and basified to pH 8 (Na₂CO₃). The yellow precipitate was filtered off, washed with water (50 ml), dried, and recrystallised from aqueous EtOH (charcoal) to give the 4,6-*diphenyl*-3-*toluidino*-2-*pyridone* (34) (261 mg, 80%) as yellow needles, m.p. 260.5—262.5 °C (Found: C, 81.8; H, 5.7; N, 7.7. C₂₄H₂₀N₂O requires C, 81.8; H, 5.7; N, 7.9%); ν_{max} . (Nujol) 3 400 (NH), 3 330, and 1 625 cm⁻¹ (C=O); δ (CDCl₃) 2.0 (3 H, s, Me), 6.5 (1 H, s, 5-H), and 6.5—8.0 (15 H, m); λ_{max} . (CHCl₃) 347 nm (ε 12 700); *m/e* (70 eV) 352 (72%), 351 (12), 335 (100), 324 (2.7), 319 (6.0), 275 (46), 220 (2.2), 217 (2.3), 204 (5.6), 191 (5.4), 176 (11), 168.5 (9.0), 104 (6.8), and 77 (9.5).

The aqueous filtrate from above was combined with the washings, and the whole acidified (concentrated HCl) and extracted with CHCl₃ (3×40 ml). The extracts were combined and dried (MgSO₄) and the solvent was removed. The residue and sublimate were recrystallised from light petroleum (b.p. 60–80 °C) to give benzoic acid (113 mg, 55%), m.p. 120–123 °C.

Hydrolysis of 5-(N-Benzoyl-p-toluidino)-4,6-diphenyl-2pyridone (35).—Aqueous H_2SO_4 (70%, 10 g) and the 5benzoyltoluidinopyridone (35) (350 mg, 0.77 mmol) were heated to 150 °C for 5 h with stirring, during which time some benzoic acid sublimed. The cooled mixture was diluted with water (40 ml) and basified to pH 8 (Na₂CO₃). The precipitate was filtered off, washed with water (25 ml), dried, and recrystallised from EtOH to give 4,6diphenyl-5-toluidino-2-pyridone (36) (183 mg, 68%) as prisms, m.p. 296-299 °C (Found: C, 81.6; H, 5.5; N, 7.9. $C_{24}H_{20}N_2O$ requires C, 81.8; H, 5.7; N, 7.9%); $\nu_{max.}$ (Nujol) 3 440, 3 320, 1 650 cm⁻¹ (C=O); δ (CDCl₃) 2.0 (3 H, s, Me), 3.2 (2 H, broad, NH₂), 6.5 (5 H, m), and 7.2 (10 H, m); δ[(CD₃)₂SO] 1.9 (3 H, s, Me), 3.6 (2 H, broad, NH_2), 6.4 [5 H, AB q, (J 10 Hz) and s], and 7.2 (10 H, m); λ_{max} (CHCl₃) 318 nm (ε 9 000); m/e (70 eV) 352 (100%), 351 (5.6), 323 (2.2), 311 (3.8), 308 (21), 307 (35), 275 (13), 221 (1.9), 206 (3.2), 104 (3.0), and 77 (3.3).

The aqueous filtrate from above and washings were combined, acidified (concentrated HCl), and extracted with $CHCl_3$ (3 × 40 ml). The extracts were combined and dried (MgSO₄), and the solvent removed. The residue and sublimate were recrystallised from light petroleum (b.p. 60-80 °C) to give benzoic acid (85 mg, 91%), m.p. 120-123 °C.

5-(N-Benzoyl-p-toluidino)-2-chloro-4, 6-diphenylpyridine (37).— 5-(N-Benzoyl-p-toluidino-4, 6-diphenyl-2-pyridone (35) (0.36 g, 0.79 mmol) in phosphoryl chloride (3 ml) was heated in a sealed Carius tube for 4 h at 195 °C and poured slowly into ice-water (150 g, 50 ml) containing sufficient aqueous NH₄OH to maintain basicity (phenolphthalein). The precipitate was filtered off, washed with water (50 ml), dried, and recrystallised from aqueous EtOH (charcoal) to give the 5-benzoyltoluidino-2-chloropyridine (37) (118 mg, 32%) as yellow needles, m.p. 172—176 °C (Found: C, 78.4; H, 4.8; N, 6.0. C₃₁H₂₃ClN₂O requires C, 78.4; H, 4.9; N, 5.9%); ν_{max} (Nujol) 1 670 cm⁻¹ (C=O); δ (CDCl₃) 2.1 (3 H, s, Me), 6.8 (1 H, s), and 7.0—7.8 (19 H, m).

3-(N-Benzoyl-p-toluidino)-2,4-diphenylpyridine (38). Pd-C (5%, 86 mg), 5-(N-benzoyl-p-toluidino)-2-chloro-4,6-diphenylpyridine (37) (110 mg, 0.28 mmol), KOH (39 mg), and MeOH (AnalaR, 25 ml) were stirred under hydrogen at 20 °C and 760 mmHg for 18 h. The reaction mixture

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was filtered and the solvent removed from the filtrate. Water (40 ml) was added to the residue and this was extracted with $ext{CHCl}_3$ (3 imes 25 ml). The extracts were combined and dried (MgSO₄) and the solvent was removed to leave a yellow solid (92 mg, 90%), m.p. 170-180 °C. Recrystallisation from EtOH afforded the 3-benzoyltoluidino-pyridine (38) (67 mg, 66%) as pale yellow needles, m.p. 191-192 °C (Found: C, 84.6; H, 5.5; N, 6.2. C₃₁- $\rm H_{24}N_{2}O$ requires C, 84.5; H, 5.5; N, 6.4%); $\nu_{\rm max.}$ (Nujol) 1.670 cm^{-1} (C=O); δ (CDCl₃) 2.1 (3 H, s, Me), 6.9-7.5 (19 H, m), 7.7 (1 H, d, J 9 Hz), and 8.8 (1 H, d, J 5.5 Hz, 6-H).

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REFERENCES

¹ Part 59, A. R. Katritzky, A. S. Afridi, and C. A. Ramsden,

Pakistan J. Sci. Ind. Res., 1978, 21, 1. ² A. R. Katritzky, A. V. Chapman, M. J. Cook, and G. H.

^a (a) M. J. Cook, A. R. Katritzky, and G. H. Millet, *Heterocycles*, 1977, **7**, 227; (b) A. R. Katritzky, M. J. Cook, S. B. Brown, R. Cruz, G. H. Millet, and A. Anani, *J.C.S. Perkin I*, 1979, 2493.

⁴ A. R. Katritzky and J. M. Lagowski, 'Chemistry of the Heterocyclic N-Oxides ', Academic Press, New York, N.Y., 1971,

e.g., p. 288, 296. ⁵ R. A. Abramovitch and I. Shinkai, Accounts Chem. Res.,

1976, **9**, 192.

⁶ C. Kaneko, J. Pharm. Soc. Japan, 1959, 79, 428 (Chem. Abs., 1959, 53, 17126b); A. Ohta and E. Ochiai, Chem. and Pharm. Bull. (Japan), 1962, 10, 1260.

⁷ L. A. Paquette, J. Amer. Chem. Soc., 1965, 87, 1407.
⁸ L. A. Paquette, J. Amer. Chem. Soc., 1965. 87, 5186.

⁹ W. König and R. Geiger, Chem. Ber., 1973, 106, 3626.

¹⁰ E. C. Taylor, F. Kienzle, and A. McKillop, J. Org. Chem., 1970, **35**, 1672.

¹¹ J. v. Braun and W. Pinkernelle, Chem. Ber., 1934, 67, 1218. ¹² R. A. Abramovitch and G. M. Singer, J. Org. Chem., 1974,

89. 1795.

J. Biehringer and A. Busch, Chem. Ber., 1902, 35, 1964.

¹⁴ H. Furrer, Tetrahedron Letters, 1974, 2953.

¹⁵ E. C. Taylor, H. W. Altland, F. Kienzle, and A. McKillop, J. Org. Chem., 1976, 41, 24.

¹⁶ T. Cohen and J. H. Fager, J. Amer. Chem. Soc., 1965, 87, 5701; R. Bodalski and A. R. Katritzky, Tetrahedron Letters, 1968, 257; T. Cohen and G. L. Deets, J. Amer. Chem. Soc., 1972,

94, 932. ¹⁷ S. Oae, T. Kitao, and Y. Kitaoka, J. Amer. Chem. Soc., 1962, **84**, 3359.

18 S. Oae, T. Kitao, and Y. Kitaoka, Tetrahedron, 1964, 20, 2685.

¹⁹ S. Tamagaki, K. Ogino, S. Kozuka, and S. Oae, Tetrahedron, 1970, 26, 4675. ²⁰ S. Oae, S. Tamagaki, and S. Kozuka, *Tetrahedron Letters*,

1966, 1513.

²¹ J. A. Berson and T. Cohen, J. Amer. Chem. Soc., 1955, 77, 1281; S. Oae, S. Tamagaki, T. Negoro, and S. Kozuka, Tetrahedron, 1970, 26, 4051; V. J. Traynelis, K. Yamauchi, and J. P.

²¹ S. Oae, T. Kitao, and Y. Kitaoka, J. Amer. Chem. Soc., 1974, 96, 7289.
²² S. Oae, T. Kitao, and Y. Kitaoka, J. Amer. Chem. Soc.,

1962, 84, 3362.

 ²³ Y. Kitaoka and S. Oac, Tetrahedron Letters, 1975, 123.
²⁴ W. E. Parham and K. B. Sloan, Tetrahedron Letters, 1971, 1947; R. A. Abramovitch, R. B. Rogers, and G. M. Singer, J.

Org. Chem., 1975, 40, 41. ²⁵ R. A. Abramovitch, A. L. Miller, T. A. Radzikowska, and

P. Tomasik, J. Org. Chem., 1979, 44, 464. ²⁶ M. Araki, Y. Kawazoe, and C. Nagata, Chem. and Pharm.

Bull. (Japan), 1969, 17, 1344. ²⁷ T. Kosuge, H. Zenda, and Y. Suzuki, Chem. and Pharm. Bull. (Japan), 1970, 18, 1068.

28 K. Ogino, S. Kozuka, and S. Oae, Tetrahedron Letters, 1969.

3559; K. Ogino and S. Oae, *Tetrahedron*, 1971, **27**, 6037. ²⁹ C. Walling and R. T. Clark, J. Amer. Chem. Soc., 1974, **96**,

4530. ³⁰ 'Dictionary of Organic Compounds,' 4th edn., Eyre and Spottiswoode, London, 1965, vol. 4, p. 2666.

³¹ W. Davies and W. H. Perkin jun., J. Chem. Soc., 1922, 121, 2202.

³² Ref. 30, vol. 5, p. 3075.

33 M. Colonna and A. Risaliti, Ann. Chim. (Italy), 1958, 48, 1395 (Chem. Abs., 1959, 53, 14683e).

³⁴ I. E.-S. El-Kholy, F. K. Rafla, and M. M. Mishrikey, J. Chem. Soc. (C), 1970, 1578.

 ³⁵ Ref. 30, vol. 1, p. 344.
³⁶ Ref. 30, vol. 5, p. 3074.
³⁷ P. Kattwinkel and R. Wolffenstein, *Chem. Ber.*, 1904, 37, 3221.

³⁸ A. I. Vogel, 'A Textbook of Practical Organic Chemistry including Qualitative Organic Analysis,' 3rd edn., Longmans, London, 1956, p. 583.

39 H. Apitzsch, Chem. Ber., 1900, 33, 3521.